SOME ASPECTS CONCERNING A NEW POSSIBILITY OF Al-SiCp COMPOSITE PRODUCTION

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The paper presents the calculations regarding the acceleration and the shape factors for aluminum and aluminum alloy-silicon carbide particle composites. At the same time, due to poor wettability between the silicon carbide and aluminum melt, a new method that uses gravity casting for Al-SiCp sieves is presented. Different mesh numbers of sieves made from bonded textile fibers and silicon carbide powder were used. Finally, a microscope analysis was carried out for all samples.

Keywords: Aluminum, silicon carbide particles, critical acceleration, shape coefficient, sieves

1. Introduction

Metal matrix composites have been developed and applied as structural materials in the aerospace and automobile industry for many decades due to their high specific strength, high specific stiffness, good high-temperature properties and better wear resistance [1].

Particle reinforced metal matrix composites are now being produced commercially, the different types of reinforcement being used together with alternative processing methods [2]. Aluminum is one of the most desirable metals owing to its desirable properties of corrosion resistance, density ($\rho=2700$ kg/m$^3$), Young modulus (70,000 MPa), tensile strength (90 MPa), and thermal expansion coefficient ($23.5\cdot10^{-6}$ /ºC) [3]. Silicon carbide is a covalent material of great technological interest due to its excellent overall properties including low density ($\rho=3.1...3.21$ g/cm$^3$), stability at high temperatures, high hardness (9.2...9.3 on the Mohs' scale), and high value for the modulus of elasticity ($E = 400,000$ MPa), low thermal expansion coefficient ($\alpha_{SiC}=4\cdot10^{-6}$ /ºC) [4-9].

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Aluminum-silicon carbide combines the benefits of high thermal conductivity of metal and low coefficient of thermal expansion of ceramic, possessing high modulus, strength values, wear resistance, high thermal stability, low weight and a more effective load carrying capacity compared to many other materials [10].

The main methods of producing aluminum metal matrix composites are: stir casting, squeeze casting, rheocasting (compo casting), and spray deposition [11-12].

2. Theoretical consideration

Although gravitational casting seems to be the simplest and cheapest method to produce aluminium-silicon carbide particles, a major problem arises concerning the achievement of the mixture by mechanical stirring due to the non-wetting conditions in the system [13, 14]. To penetrate the aluminium melt, a silicon carbide particle needs to exceed a critical acceleration [13]. For a spherical particle, the critical acceleration value can be determined by the relationship:

$$a_{cr} = -\frac{3}{2} \frac{\sigma_{lg}}{r_p^2 (\rho_p - \rho_l)} \cdot \cos \theta$$  \hspace{1cm} (1)

where: $\sigma_{lg}$ - the surface tension at the aluminium melt-gas interface, $\cos \theta$ - the wetting degree, $r_p$ - the radius of the particle, $\rho_p$ - the particle density; $\rho_l$ - the density of the liquid aluminium.

For a spherical particle, equation (1) becomes:

$$r_p = \left[\frac{3m_p}{4\pi \rho_p^3}\right]$$  \hspace{1cm} (2)

$$a_{cr} = k_v \left[\frac{\rho_p}{m_p}\right]^{\frac{3}{2}} \frac{\sigma_{lg} \cos \theta}{\rho_l - \rho_p}$$  \hspace{1cm} (3)

where: $k_v$ is a volumetric coefficient of shape, $m_p$ - the particle mass.

For a spherical particle $k_v = 3.896$.

The calculated values of the volumetric coefficient shape for different types of particles are presented in Table 1.
Table 1

Calculated values of volumetric coefficient shape

<table>
<thead>
<tr>
<th>Shape of particles</th>
<th>Mathematical formula of $K_v$</th>
<th>Value of $K_v$</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sphere, $r = r_p$</td>
<td>$(6\pi^2)^{\frac{1}{3}}$</td>
<td>3.896</td>
</tr>
<tr>
<td>Cylinder, $r = r_p$, $h = 2r_p$</td>
<td>$3\left[\frac{2\gamma}{\pi}\right]^{\frac{1}{3}}$</td>
<td>5.106</td>
</tr>
<tr>
<td>Conical, $r = r_p$, $h = 2r_p$</td>
<td>$\frac{\sqrt{5} + 1}{2} \cdot \left[\frac{3\pi^{\frac{1}{3}}}{2}\right]^{\frac{1}{3}}$</td>
<td>3.972</td>
</tr>
<tr>
<td>Conical frustum, $R = r_p$, $r = \frac{R}{2}$, $h = 2r_p$</td>
<td>$5 + 3\sqrt{17}\left[\frac{3\pi^{\frac{1}{3}}}{2^{\frac{2}{3}}}</td>
<td></td>
</tr>
</tbody>
</table><p>ight]^{\frac{1}{3}}$ | 4.423          |
| Cube, $l = l_p$              | $6 + \sqrt{3}$               | 6              |
| Prism triangle, $l = l_p$, $h = l_p$ | $\frac{6 + \sqrt{3}}{(2\sqrt{3})^{\frac{1}{3}}}$ | 5.114          |
| Prism rectangle, $L = l_p$, $l = \frac{L}{2}$, $h = l_p$ | $\frac{7}{2^{\frac{3}{2}}}$ | 5.04           |
| Pyramid quadrate, $l = l_p$, $h = l_p$ | $\frac{3^{\frac{1}{3}}(\sqrt{5} + 1)}{2^{\frac{1}{3}}}$ | 4.667          |
| Regular pyramid (equilateral triangle base), $l = l_p$, $h = l_p$ | $(\sqrt{3^3} + \sqrt{3}) \cdot (\frac{\sqrt{3}}{2^4})^{\frac{1}{3}}$ | 3.802          |
| Pyramid (hexagonal base), $l = l_p$, $h = l_p$ | $\frac{3^{\frac{1}{2}}}{2^{\frac{3}{2}}} \cdot (\sqrt{7} + \sqrt{3})$ | 6.889          |
| Truncated square pyramid, $L = l_p$, $l = \frac{L}{2}$, $h = l_p$ | $\frac{3\sqrt{17} + 5 \cdot 12^{\frac{1}{3}}}{4 \cdot 7}$ | 5.197          |</p>

Fig. 1 and Fig. 2 present the critical acceleration values for particles of different shapes. The calculations were made for 973 K, and a value of contact angle for oxidised particles of SiC [15], Shen and Keene’s relationships for surface tension of aluminium melt, Luca’s equation for the density of aluminium liquid and a particle mass of $m_p = 1.344 \cdot 10^{-8}$ kg (equivalent to a spherical particle with a radius of 100 µm) [16, 17].
Fig. 1. Critical acceleration for round particles at 973K

Fig. 2. Critical acceleration for polyhedral particles at 973K

From Fig. 1 and Fig. 2, the results show that triangle particles have minimum critical acceleration values compared to other shapes, but in spite of that, they are not sufficient for particle penetration.

The high values obtained by calculating the critical acceleration of particles show the need to apply a number of measures to improve the wetting conditions. The recommended measures, which include alloying of the melt with surface active elements, overheating of the metallic bath, and coating the particles or heat treatment of the complementary material, do not lead to a significant change in the system conditions.

This work presents a new method to obtain aluminium – silicon carbide particle composites by gravitational casting to improve the degree of incorporation between silicon carbide and melt aluminium.
3. Experimental conditions

Materials

Commercial aluminum was employed in this study; its chemical composition is listed in Table 2 and the major alloying element was silicon.

<table>
<thead>
<tr>
<th>Elements</th>
<th>Al</th>
<th>Fe</th>
<th>Cu</th>
<th>Mn</th>
<th>Mg</th>
<th>Si</th>
<th>Zn</th>
<th>Others</th>
</tr>
</thead>
<tbody>
<tr>
<td>Wt. %</td>
<td>97.95</td>
<td>0.574</td>
<td>0.165</td>
<td>0.107</td>
<td>0.212</td>
<td>0.731</td>
<td>0.187</td>
<td>0.076</td>
</tr>
</tbody>
</table>

Table 2

Sieves made from bonded textile fibers and SiC powders were used as raw materials. Four different sizes of SiCp sieves: 40 mesh, 120 mesh, 180 mesh, and 220 mesh were chosen as the reinforcements.

Procedure

Sand molds were prepared and heated to 325°C to remove any moisture and reduce the temperature difference between the mold and the melt. The bonded SiCp sieve covered both the base and inside wall of the mold, as shown in Fig. 3. The thicknesses of these sieves were 1.13 mm (for 40 mesh), 0.72 mm (120 mesh), 0.67 mm (180 mesh), 0.6 mm (220 mesh).

Fig. 3. Section of sand mold padded in SiCp

Al melt was prepared in a gas furnace, using a graphite crucible. Pouring aluminum melt in the mold was carried out at 900°C. After solidification, the samples were sectioned with an abrasive cutting machine into samples of 15 mm thickness from the middle, without causing any damage to the casting surface. Samples were ground with 120, 180, 320, 600, 800, 1000 grit papers respectively. Finally the polishing was finished on cloth using a diamond paste solution of 9
μm, 3 μm respectively, 2 min for each solution. The particle distribution and presence of SiC<sub>p</sub> in the cast composites was identified by means of an optical microscope. The optical microscope is used for quality control applications and detailed examination of newly developed materials, metals, and chemicals. In this work, the instrument was used to study the microstructure of Al-SiC<sub>p</sub> composite samples obtained by different methods, followed by immage analysis.

4. Results and discussions

Fig. 4 (a-d) shows the pictures of the silicon carbide powder from sieves of different mesh sizes.

![Fig. 4. Silicon carbide powder (a-particle size =0.587mm, b- particle size =0.302mm, c- particle size =0.276mm, d- particle size =0.251mm).](image)

Table 3 list the size of SiC<sub>p</sub> particles which were obtained from the optical microscope analysis.

<table>
<thead>
<tr>
<th>Mesh no.of sieve</th>
<th>Mesh 40</th>
<th>Mesh 120</th>
<th>Mesh 180</th>
<th>Mesh 220</th>
</tr>
</thead>
<tbody>
<tr>
<td>Particle size (mm)</td>
<td>0.587</td>
<td>0.302</td>
<td>0.276</td>
<td>0.251</td>
</tr>
</tbody>
</table>

Table 3

The influence of particle shape on their behavior in the melt was investigated.

For the morphological characterization of SiC particles, different shape coefficients were measured; elongation \((EL = F/R)\), \(F\) is the line connecting the two most distant points on the circumference, \(R\) - the equivalent rectangular shortest side; sphericity \((\Psi_w = d_v^2/d_s^2)\), \(d_v\) and \(d_s\) are the equivalent volume and surface diameter respectively; shape factor \((K_j = \text{thickness} / \sqrt{\text{length} \cdot \text{width}})\) [18-20].
The mean values of the shape coefficients of the particles were calculated (Fig. 4) and are listed in (Table 4).

<table>
<thead>
<tr>
<th>Particle size (mm)</th>
<th>Elongation</th>
<th>Shape factor</th>
<th>Sphericity</th>
<th>Aspect ratio</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.587</td>
<td>1.287</td>
<td>0.554</td>
<td>0.429</td>
<td>1.4</td>
</tr>
<tr>
<td>0.302</td>
<td>1.585</td>
<td>0.573</td>
<td>0.45</td>
<td>1.54</td>
</tr>
<tr>
<td>0.276</td>
<td>1.59</td>
<td>0.601</td>
<td>0.461</td>
<td>1.553</td>
</tr>
<tr>
<td>0.251</td>
<td>1.644</td>
<td>0.603</td>
<td>0.479</td>
<td>1.58</td>
</tr>
</tbody>
</table>

When SiC_p was added into the aluminum melt, a number of modifications pertaining to its incorporation were observed, that will cause changes in its features, such as the mean shape coefficients’ value, as a result of exposure to the phenomenon of erosion or the addition of impurities. The optical micrographs of aluminum composites reinforced with sieves of SiC_p are shown in Fig. 5.

![Microstructure of Al-SiC_p composite for different sizes of powder sieves reinforced with aluminum: particle size a=0.587mm, b =0.302mm, c =0.276mm, d=0.251mm).](image)

Very irregular shapes and particle distribution was obtained with a clear tendency of inter granular distribution of the SiC particles. This may be caused by the low inoculation capacity of the SiC particles, depending of their inadequate surface properties or by the too low cooling rate of the melt during solidification.

Large size silicon carbide particles tend to segregate at the crystal grain boundaries. This segregation during solidification is harder to control, but can be limited by using a copper plate under the sand mold to increase the cooling rate and conduct crystallization. A superficial layer of Al-SiC_p composite improves the
tribological properties of moving and contacting components, such as the wear resistance in some automobile parts.

As seen in the results shown in Fig. 6, there seems to be an optimum size for SiC particles (0.276 mm) which assures a higher particle density on the sample section. This could be explained by the fact that the settlement of particles in the melt is already starting to occur when the melt is still in the mould. This is because the density of SiC<sub>p</sub> is higher than that of the molten aluminum, which leads to settling or sedimentation of the particle reinforcements; therefore the higher sedimentation tendency of the bigger particles (0.587 mm, 0.302 mm) occurs when the alloy is in a liquid alloy state. At the same time, the smaller particles (0.251 mm) will tend to behave similarly to the bigger particles due to the agglomeration tendency, because

\[ \Delta E_x = \sigma_1 \cos \theta, \quad \theta > 0 \text{ and } \Delta E_x < 0. \]

![Fig. 6. Effect of particle size on silicon carbide particle density in the aluminum matrix (Mesh 40= particle size 0.587mm, Mesh 120= particle size 0.302mm, Mesh 180= particle size 0.276mm, Mesh 20= particle size 0.251mm)](image)

Although there are significant difficulties in the quantification of particle shape, especially for irregular particles, the analysis of the several parameters proposed so far to describe both the SiC<sub>p</sub> and Al-embedded SiC particles shapes has been generally accepted [21]. In this content, the results of shape coefficients’ analysis, both of individual SiC particles (SiC powder) and Al-embedded SiC particles, are presented and compared in Fig.s 7-10.
As can be seen from Figs. 7-10, the values of SiC particle shape coefficients are constantly higher than the similar values for Al-embedded SiC particles, with a more significant difference for the mean of the sphericity factor (Fig. 8) [22]. This could be explained by a hollow which occurs around the free individual SiC particles, compared with the every sharp image of the embedded particles in the Al matrix.
Fig. 9. Effect of particle size on silicon carbide mean elongation in the aluminum matrix (Mesh 40 = particle size 0.587mm, Mesh 120 = particle size 0.302mm, Mesh 180 = particle size 0.276mm, Mesh 220 = particle size 0.251mm)

Fig. 10. Effect of particle size on silicon carbide mean shape factor in the aluminum matrix (Mesh 40 = particle size 0.587mm, Mesh 120 = particle size 0.302mm, Mesh 180 = particle size 0.276mm, Mesh 220 = particle size 0.251mm)

It was found that the elongation and aspect ratio of SiC\(_p\) were much higher than for Al-embedded SiC\(_p\); this may be explained by a chemical reaction possibly occurring between the SiC particles and the melt aluminum alloy that forms another compound which remains in the melt, such:

\[4\text{Al} + 3\text{SiC} \rightarrow \text{Al}_4\text{C}_3 + 3\text{Si}\]

The continuous reaction layer of Al\(_4\)C\(_3\) occurred at 900\(^\circ\)C between the melt aluminum and the silicon carbide [23]. The Al\(_4\)C\(_3\) forms at the interface, while the Si dissolves in the Al matrix and causes a reduction in the shape coefficients’ values.
5. Conclusions

A number of important conclusions may be drawn, as follows:

- Particle shape was an important parameter affecting the volumetric coefficient, and thus having an effect on the critical acceleration at different percentages (critical acceleration of conical frustum particle 62% higher than that of the prism triangle particle).

Due to the high acceleration values necessary for SiC<sub>p</sub> to penetrate into the melt aluminum, a new method (sieves made from bonded SiC<sub>p</sub> textile fibers) was used.

- There seems to exist an optimum SiC particle size (0.276mm) which promotes a higher particle density on the cast sample section (10.7 no. of particle/µm<sup>2</sup>).

- Sphericity, elongation, shape factor and aspect ratio for SiC<sub>p</sub> were higher than for the Al-SiC<sub>p</sub> composite, due to the chemical reaction which took place between the SiC<sub>p</sub> surface and aluminum. In addition, a number of defects, such as a hollow, led to a reduction in shape factors for Al-SiC<sub>p</sub>.

- The segregation of silicon carbide particles must be slowed by high-speed solidification to improve the incorporation of the particles and the melt aluminum alloy.

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